

アルコール燃料の燃焼排出物の 生成経路

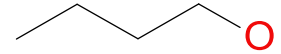
防衛大学校応用化学科 安永健治

Previous combustion studies of butanol isomers

Spark ignition engine

- R. W. Rice, A. K. Sanyal, A. C. Elrod, J. Eng. Gas Power 113 (1991) 377-382
T. Wallner, S. A. Miers, S. McConnell, J. Eng. Gas Power 131 (2009) 32802-32809
G. L. Miller, J. L. Smith, J. P. Workman, Trans. ASAE 24 (1981) 538-540
M. Gautam, D. W. Martin, Proc. Inst, Mech. Eng. Part A 214 (2000) 165-182
M. Gautam, D. W. Martin, D. Carder, Proc. Inst, Mech. Eng. Part A 214 (2000) 497-511
Y. Yacoub, R. Bata, M. Gautam, Proc. Inst, Mech. Eng. Part 212 (1998) 363-379
F.N. Alasfour, Appl. Therm. Eng. 17 (1997) 537-549
F.N. Alasfour, Int. J. Energy Res. 21 (1997) 21-30
S. Szwaja, J. D. Naber, Fuel 89 (2010) 1573-1582

n-Butanol

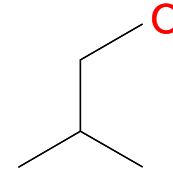


Compression ignition engine

- G. L. Miller, J. L. Smith, J. P. Workman, Trans. ASAE 24 (1981) 538-540
M. Yao, H. Wang, Z. Zheng, Y. Yue, Fuel 89 (2010) 2191-2201
C. D. Rakopoulos, A. M. Diamaratos, E. G. Giakoumis, D. C. Rakopoulos, Energy 35 (2011) 5173-5184
F. Lujaji, L. Kristóf, A. Bereczky, M. Mbarawa, Fuel 90 (2011) 505-510.
D. C. Rakopoulos, C. D. Rakopoulos, D. T. Hountalas, E. C. Kakaras, E. G. Giakoumis, R. G. Papagiannakis, Fuel 89 (2010) 2781-2790
D. C. Rakopoulos, C. D. Rakopoulos, E. G. Giakoumis, A. M. Diamaratos, D. C. Energ. Convers. Manage. 51 (2010) 1989-1997
O. Doğan, Fuel 90 (2011) 2467-2472.
P. Saisirirat, F. Foucher, S. Chanchaona, C. Mounaïm-Rousselle, Energ. Fuel. 24 (2010) 5404–5409.
P. Saisirirat, C. Togbé, S. Chanchaona, F. Foucher, C. Mounaïm-Rousselle, P. Dagaut, Proc. Combust. Inst. 33 (2010) 3007-3014.
S. Lebedevas, G. Lebedeva, E. Sendzikiene, V. Makareviciene, Energ. Fuel. 24 (2010) 4503–4509.
R.N. Mehta, M. Chakraborty, P. Mahanta, P.A. Parikh, Ind. Eng. Chem. Res 49 (2010) 7660–7665.

Previous combustion studies of butanol isomers

iso-Butanol



Spark ignition engine

F.N. Alasfour, Appl. Therm. Eng. 18 (1998) 245–256.

F.N. Alasfour, Appl. Therm. Eng. 18 (1998) 609–618.

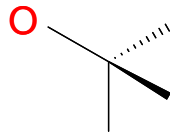
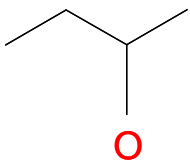
F.N. Alasfour, Energ. Source. 21 (1999) 379–394.

Compression ignition engine

M. Karabektas, M. Hosoz, Renew. Energ. 34 (2009) 1554–1559.

M.I. Al-Hasan, M. Al-Momany, Transport 23 (2008) 306–310.

sec-Butanol and *tert*-Butanol



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Gas analysis of oxidation of butanol isomers

Jet Stirred Reactor

n-Butanol

P. Dagaut, S.M. Sarathy, M.J. Thomson, Proc. Combust. Inst. 32 (2009) 229–237.

sec-Butanol, *iso*-Butanol

C. Togbè, A. Mzè-Ahmed, P. Dagaut, Energ. Fuel. 24 (2010) 5244–5256.

Premixed laminar low pressure flame

n-Butanol

N. Hansen, M. R. Harper, W. H. Green, Phys Chem. Chem. Phys. 13 (2011) 20262-20274

P. Oßwald, H. Güldenbergl, K. Kohse-Höinghaus, B. Yang, T. Yuan, F. Qi, Combust. Flame 158 (1) (2011) 2-15.

sec-Butanol, *iso*-Butanol, *tert*-Butanol

P. Oßwald, H. Güldenbergl, K. Kohse-Höinghaus, B. Yang, T. Yuan, F. Qi, Combust. Flame 158 (1) (2011) 2-15.

Flow reactor

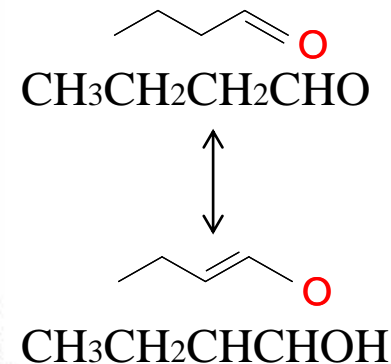
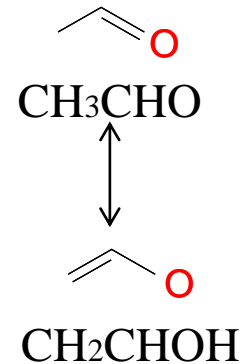
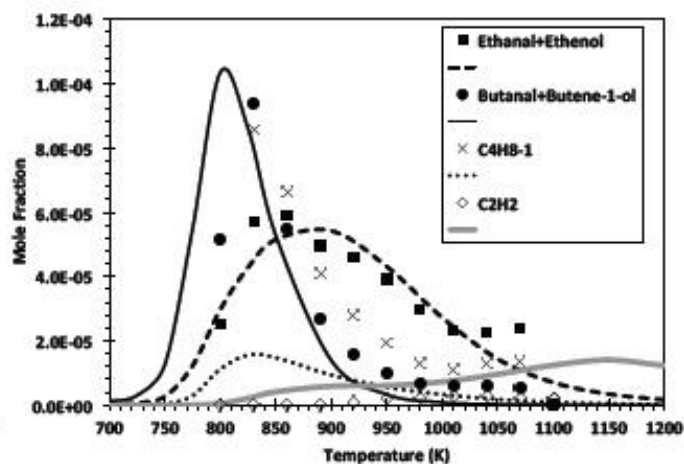
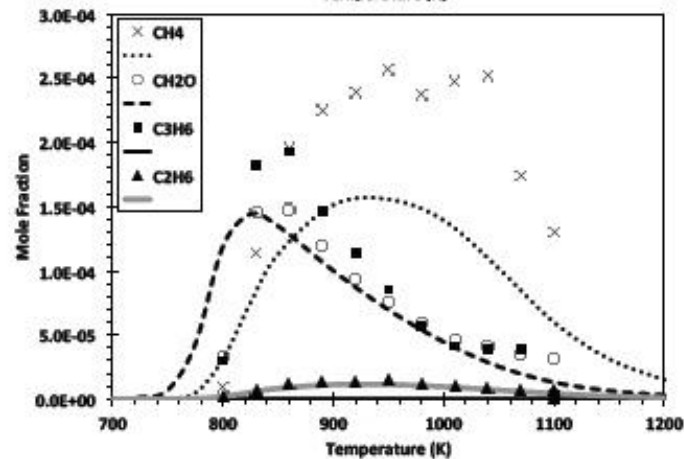
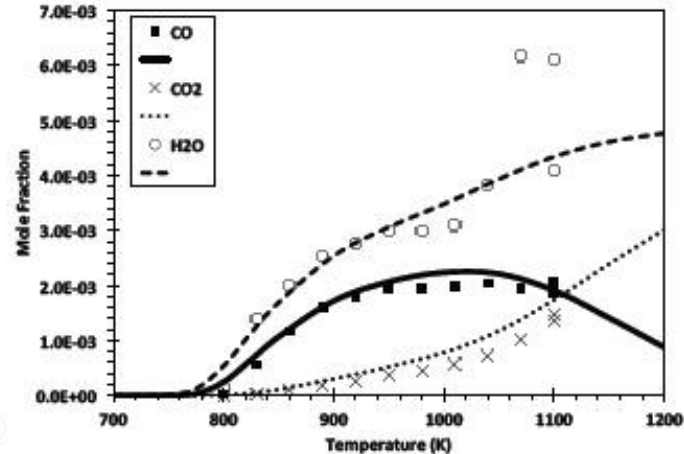
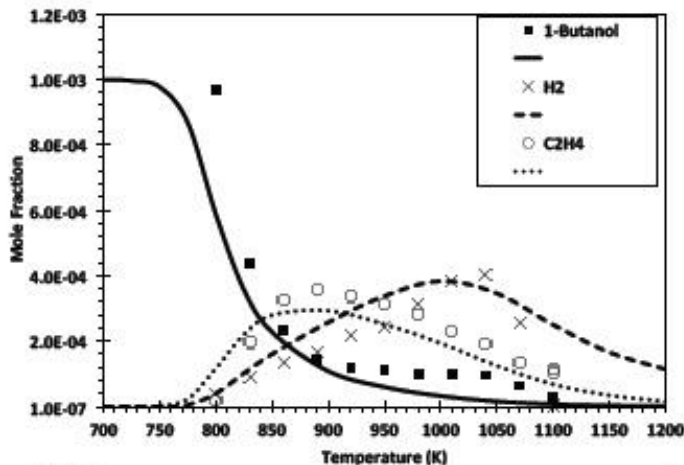
n-Butanol

M.R. Harper, K.M. Van Geem, S.P. Pyl, G.B. Marin, W.H. Green, Combust. Flame 158 (2011) 16–41.

sec-Butanol

K.M. Van Geem, S.P. Pyl, G.B. Marin, M.R. Harper, W.H. Green, Ind. & Eng. Chem. Res., 49 (2010) 10399–10420.

Jet stirred reactor



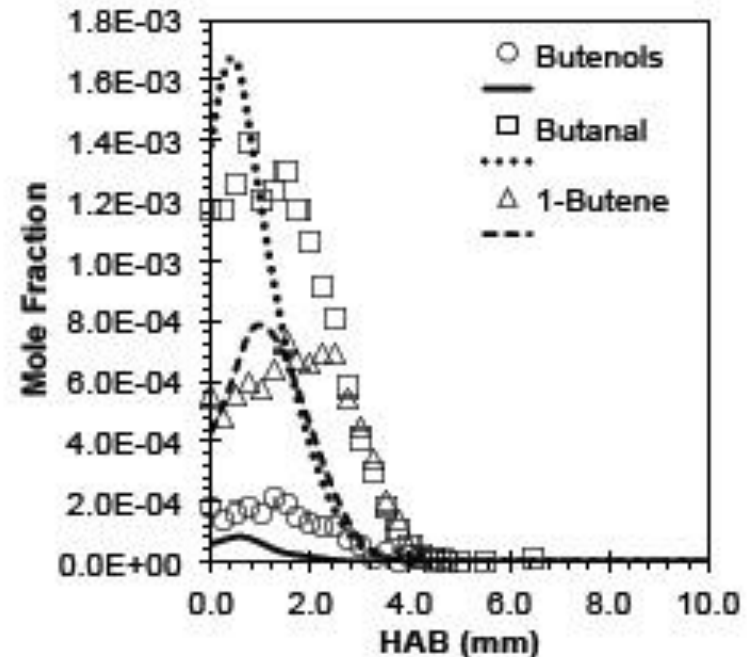
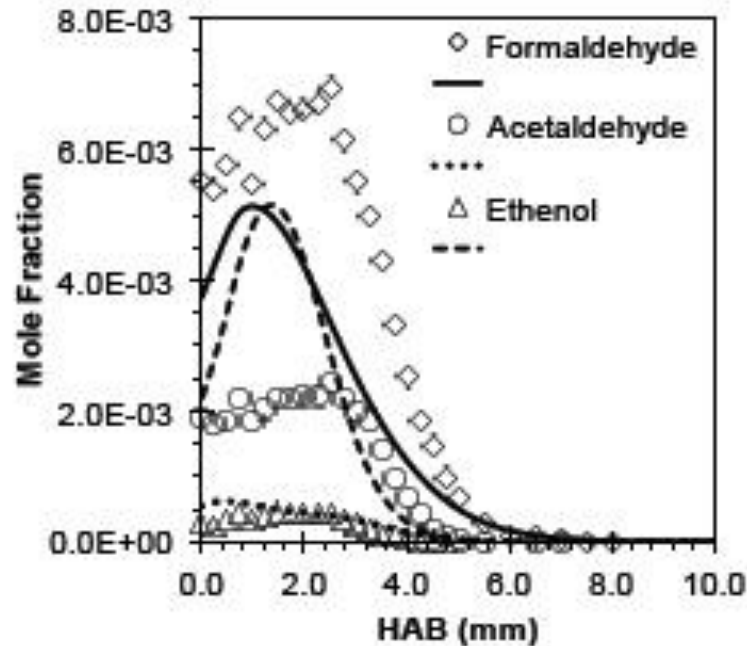
1(*n*)-butanol oxidation in a JSR at 10 atm, $\tau = 0.7$ s and $\phi = 1$. The initial fuel mole fraction was 0.1%.

P. Dagaut, S.M. Sarathy, M.J. Thomson, Proc. Combust. Inst. 32 (2009) 229–237.

Reaction Mechanism

Mani Sarathy, Stijn Vranckx, Kenji Yasunaga, Marco Mehl, Patrick Oßwald, Wayne K. Metcalfe, Charles K. Westbrook, William J. Pitz, Katharina Kohse-Höinghaus, Ravi X. Fernandes, Henry J. Curran, Combustion and Flame, accepted.

Premixed laminar low pressure flame



n-Butanol, 42.8% O₂, 50.0% Ar, $\phi = 1.0$, $p = 15$ Torr and $v = 96.1$ cm/s

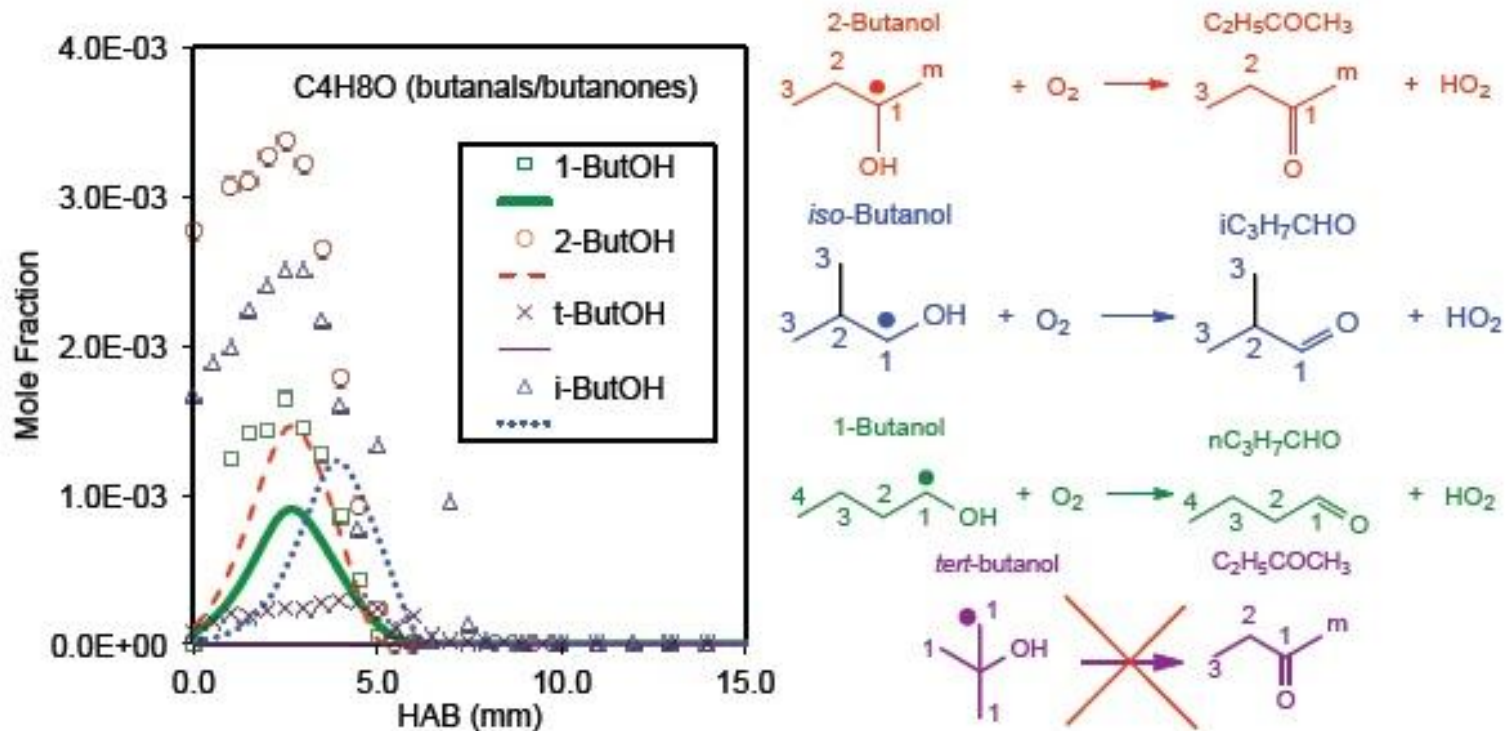
Experimental and predicted profiles for select species measured in stoichiometric 1(*n*)-butanol flames by Hansen et al.

N. Hansen, M. R. W. H. Green, Phys Chem. Chem. Phys. 13 (2011) 20262-20274

Reaction Mechanism

Mani Sarathy, Stijn Vranckx, Kenji Yasunaga, Marco Mehl, Patrick Oßwald, Wayne K. Metcalfe, Charles K. Westbrook, William J. Pitz, Katharina Kohse-Höinghaus, Ravi X. Fernandes, Henry J. Curran, Combustion and Flame, accepted.

Premixed laminar low pressure flame



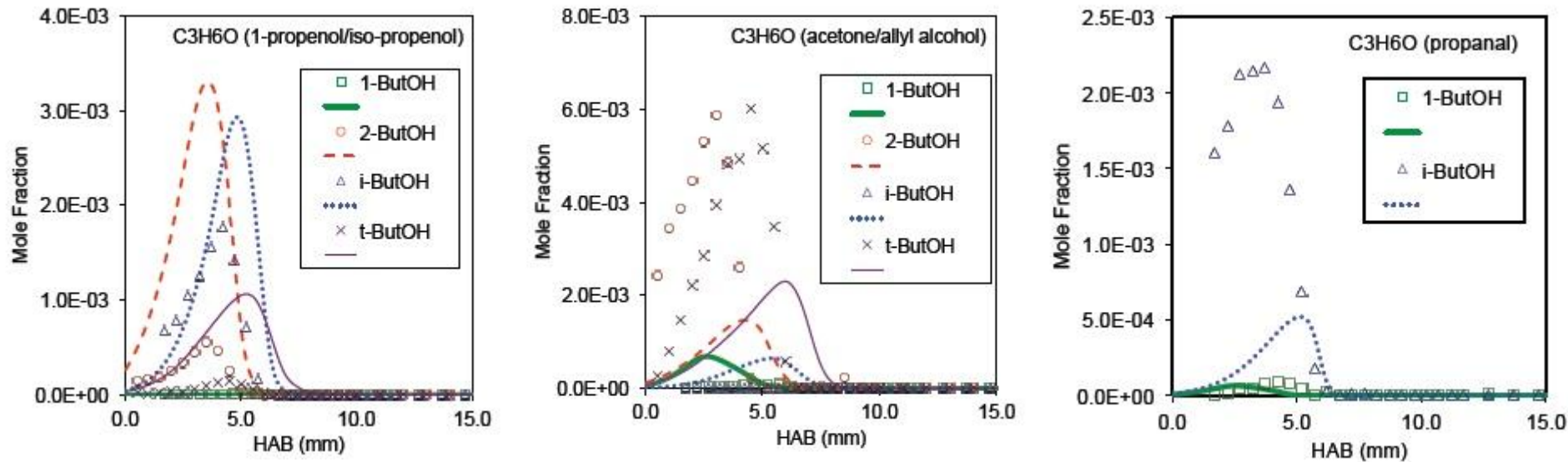
Butanol, 58.1% O₂, 25% Ar, $\phi = 1.7$, $p = 30$ Torr and $v = 32.3$ cm/s

P. Oßwald, H. Güldenber, K. Kohse-Höinghaus, B. Yang, T. Yuan, F. Qi, *Combust. Flame* 158 (1) (2011) 2-15.

Reaction Mechanism

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Premixed laminar low pressure flame



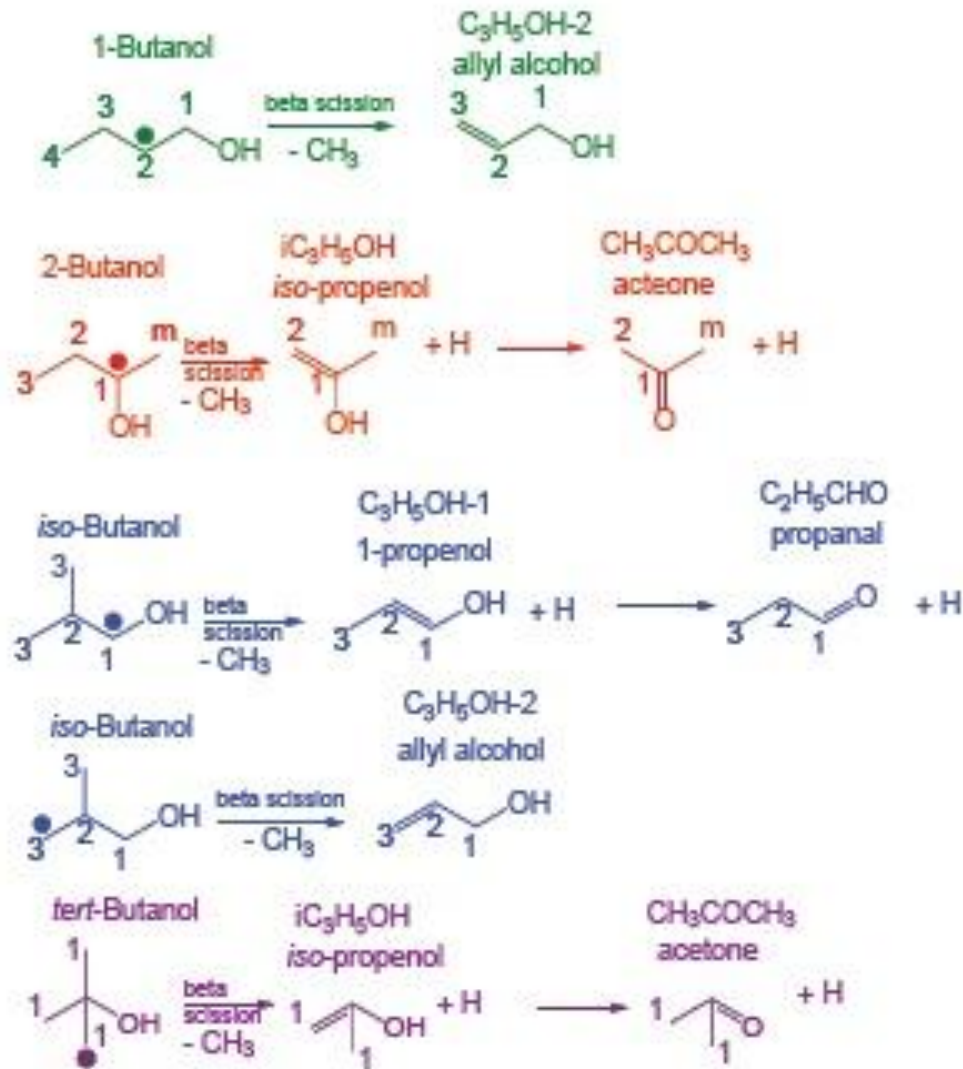
P. Oßwald, H. Güldenber, K. Kohse-Höinghaus, B. Yang, T. Yuan, F. Qi, *Combust. Flame* 158 (1) (2011) 2-15.

Reaction Mechanism

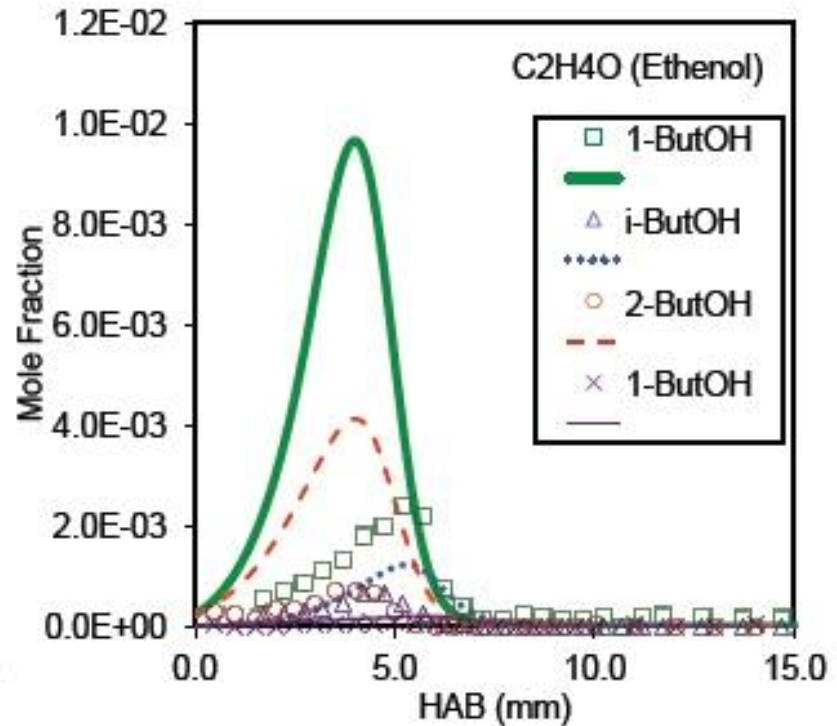
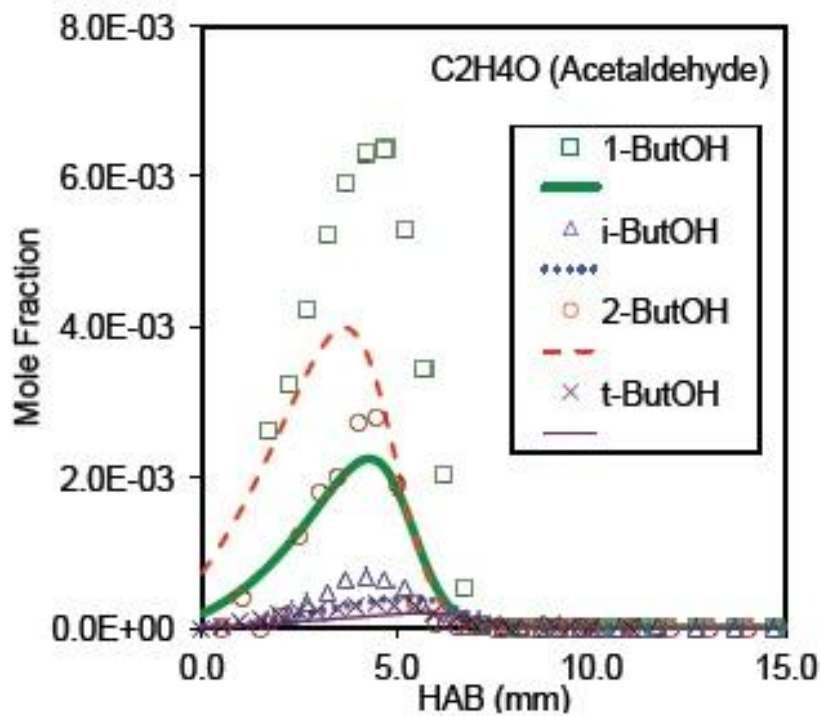
Mani Sarathy, Stijn Vranckx, Kenji Yasunaga, Marco Mehl, Patrick Oßwald, Wayne K. Metcalfe, Charles K. Westbrook, William J. Pitz, Katharina Kohse-Höinghaus, Ravi X. Fernandes, Henry J. Curran, *Combustion and Flame*, accepted.

Premixed laminar low pressure flame

Reaction Path



Premixed laminar low pressure flame

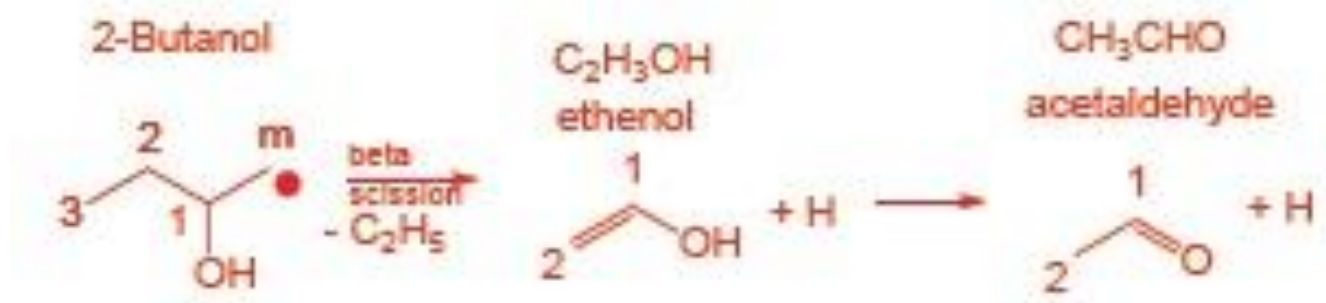
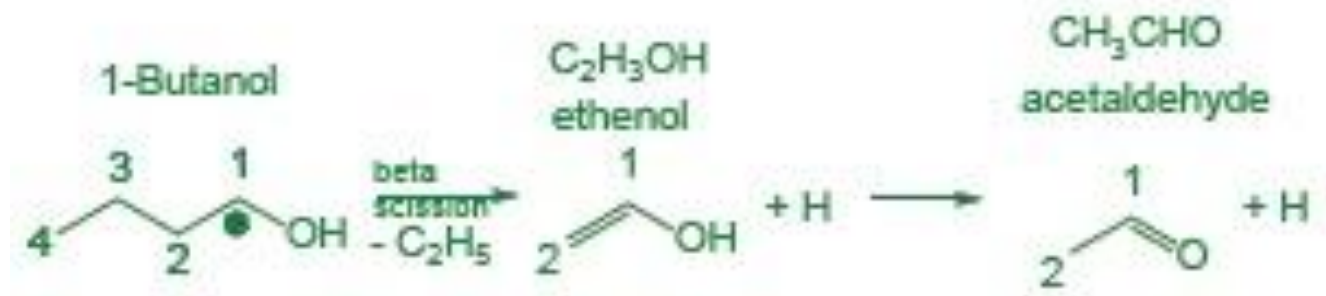


P. Oßwald, H. Güldenbergl, K. Kohse-Höinghaus, B. Yang, T. Yuan, F. Qi, Combust. Flame 158 (1) (2011) 2-15.

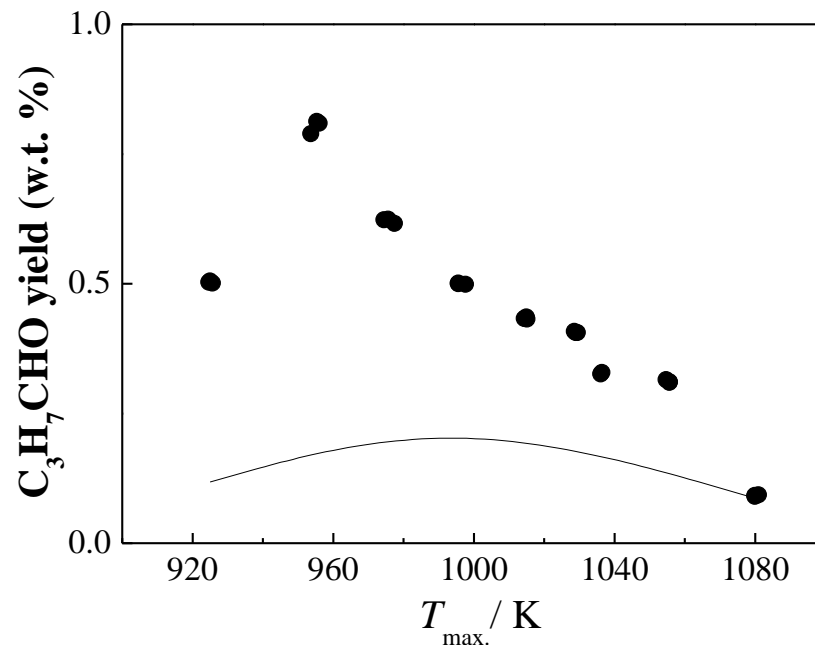
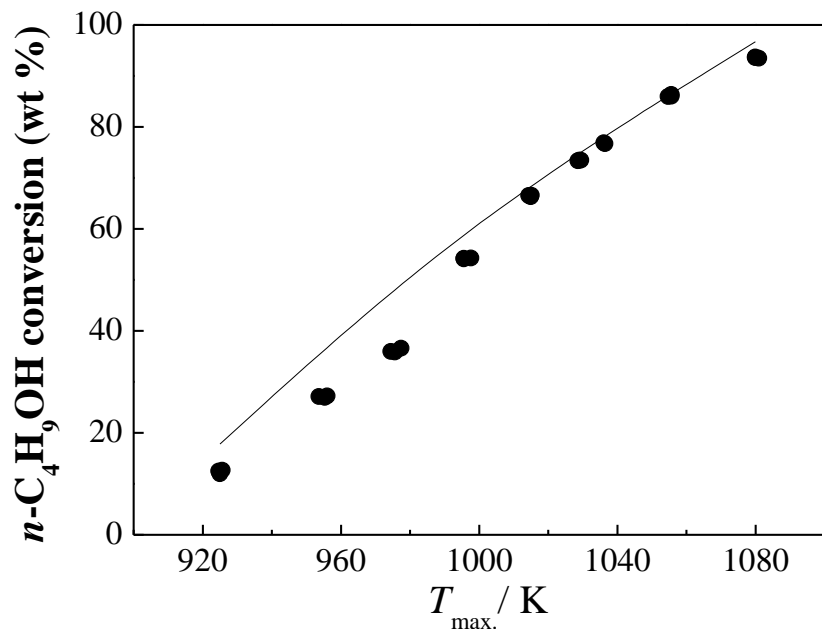
Reaction Mechanism

Mani Sarathy, Stijn Vranckx, Kenji Yasunaga, Marco Mehl, Patrick Oßwald, Wayne K. Metcalfe, Charles K. Westbrook, William J. Pitz, Katharina Kohse-Höinghaus, Ravi X. Fernandes, Henry J. Curran, Combustion and Flame, accepted.

Premixed laminar low pressure flame Reaction Path



Flow Reactor



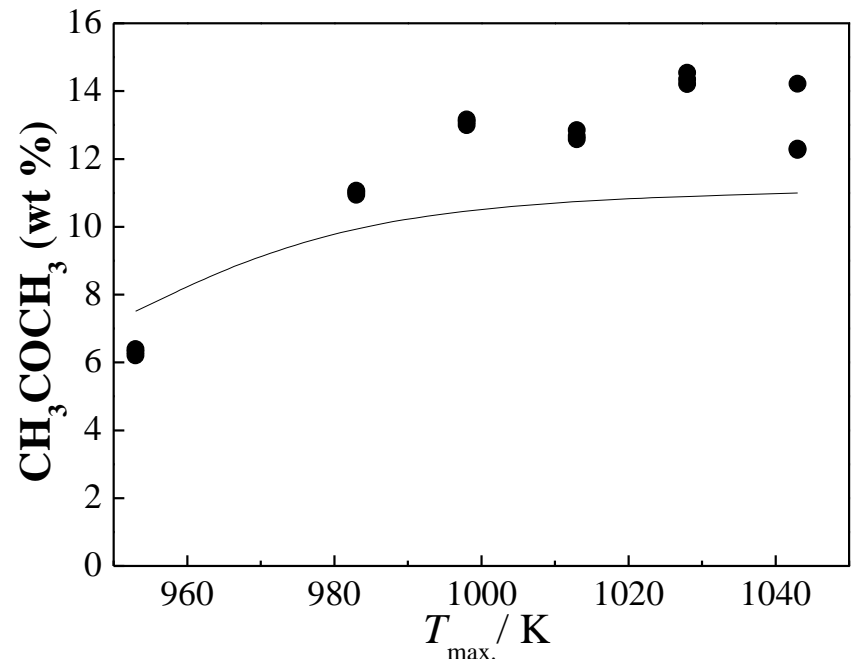
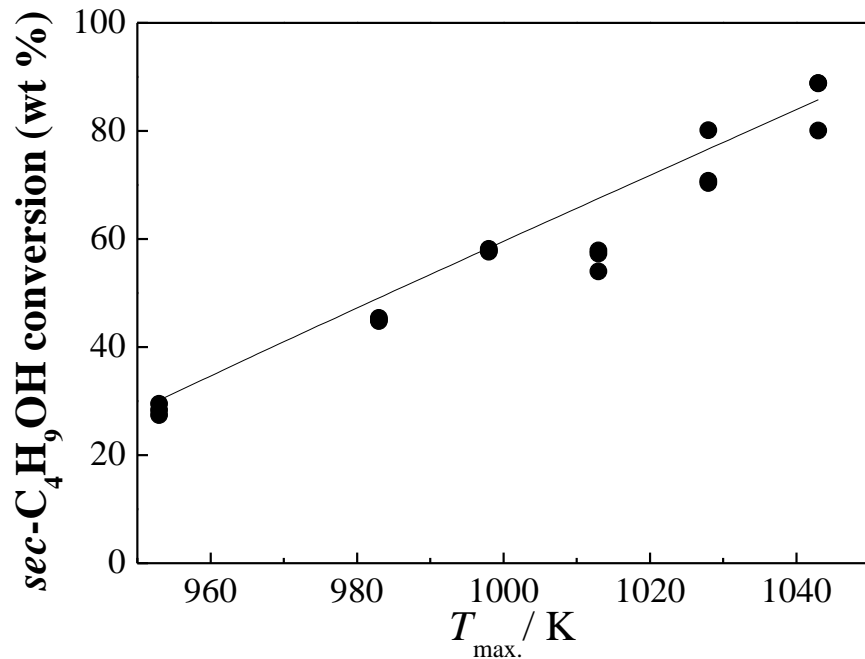
Conversion and yield profiles for $n\text{-C}_4\text{H}_9\text{OH}$ and n -butyraldehyde ($\text{C}_3\text{H}_7\text{CHO}$) in the pyrolysis reactor at 1.7 atm.

M.R. Harper, K.M. Van Geem, S.P. Pyl, G.B. Marin, W.H. Green, *Combust. Flame* 158 (2011) 16–41.

Reaction Mechanism

Mani Sarathy, Stijn Vranckx, Kenji Yasunaga, Marco Mehl, Patrick Oßwald, Wayne K. Metcalfe, Charles K. Westbrook, William J. Pitz, Katharina Kohse-Höinghaus, Ravi X. Fernandes, Henry J. Curran, *Combustion and Flame*, accepted.

Flow Reactor



Conversion and yield profiles for *sec*-C₄H₉OH and *acetone* in the pyrolysis reactor at 1.7 atm measured by Van Geem *et al.*

K.M. Van Geem, S.P. Pyl, G.B. Marin, M.R. Harper, W.H. Green, *Ind. & Eng. Chem. Res.*, 49 (2010) 10399–10420.

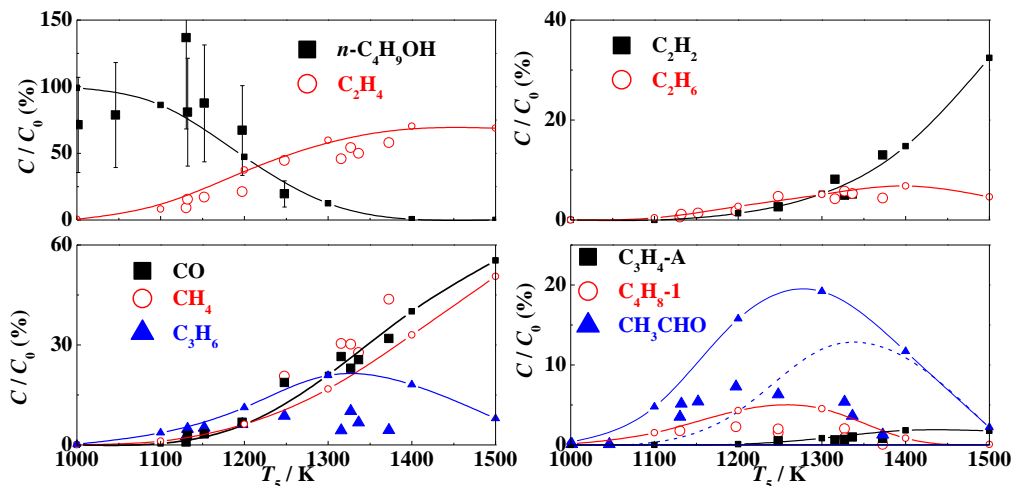
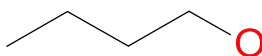
Reaction Mechanism

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Shock tube

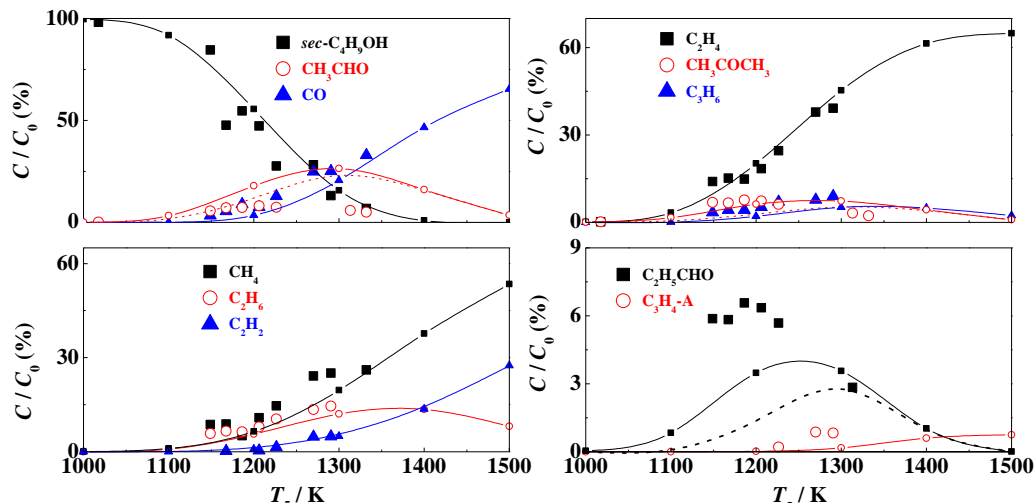
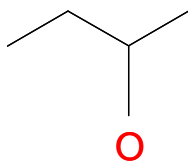
Shock tube

n-Butanol

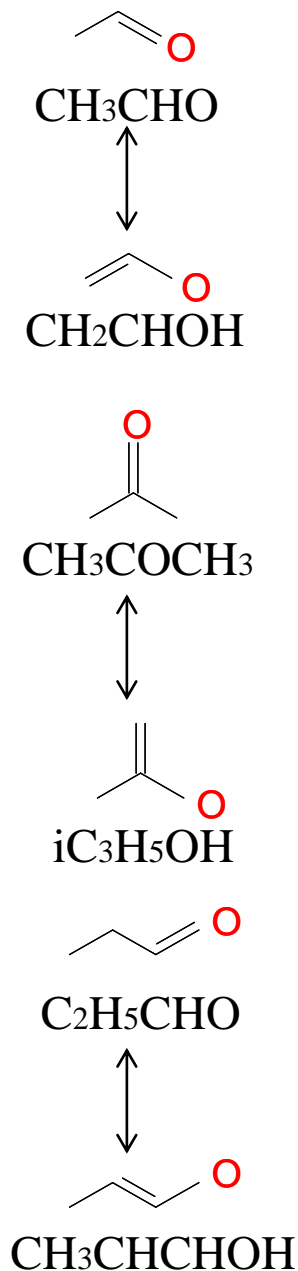


Normalized species concentration profiles for the pyrolysis of 1.5% *n*-C₄H₉OH diluted in Ar, effective heating time = 2.6–3.0 ms, lines simulation.

sec-Butanol



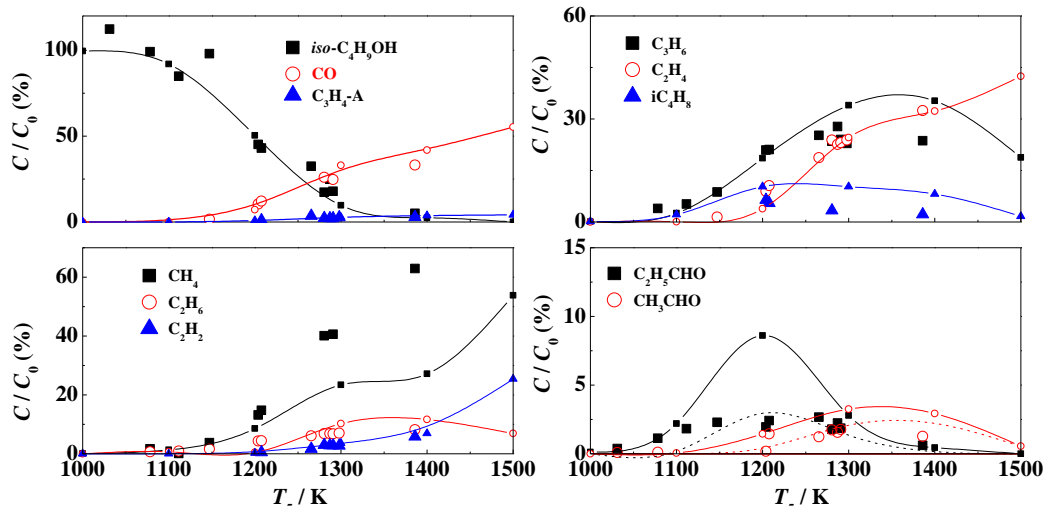
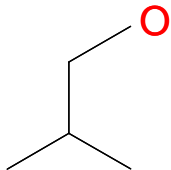
Normalized species concentration profiles for the pyrolysis of 1.5% *sec*-C₄H₉OH in Ar, effective heating time = 2.6–3.0 ms, lines simulation.



Shock tube

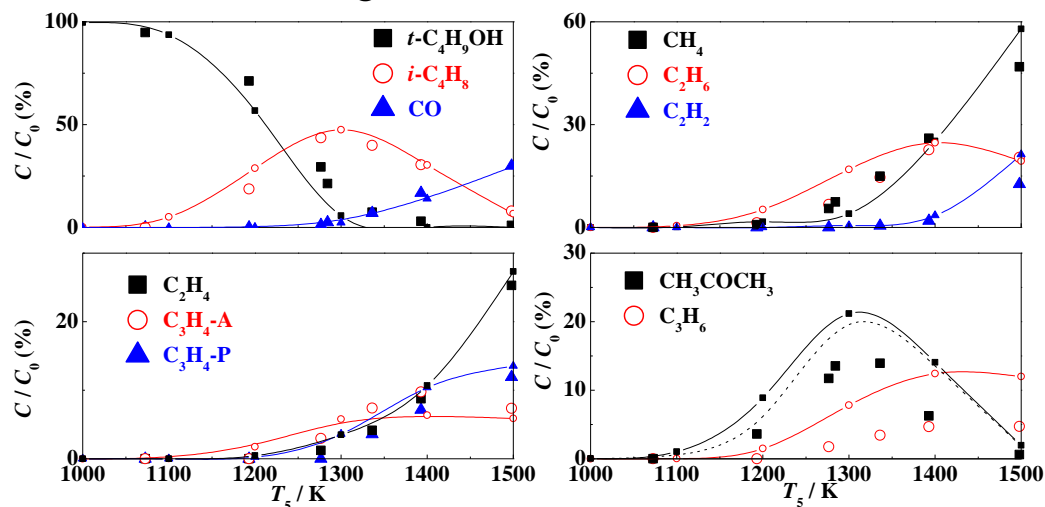
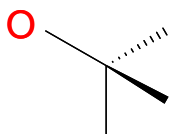
Shock tube

iso-Butanol



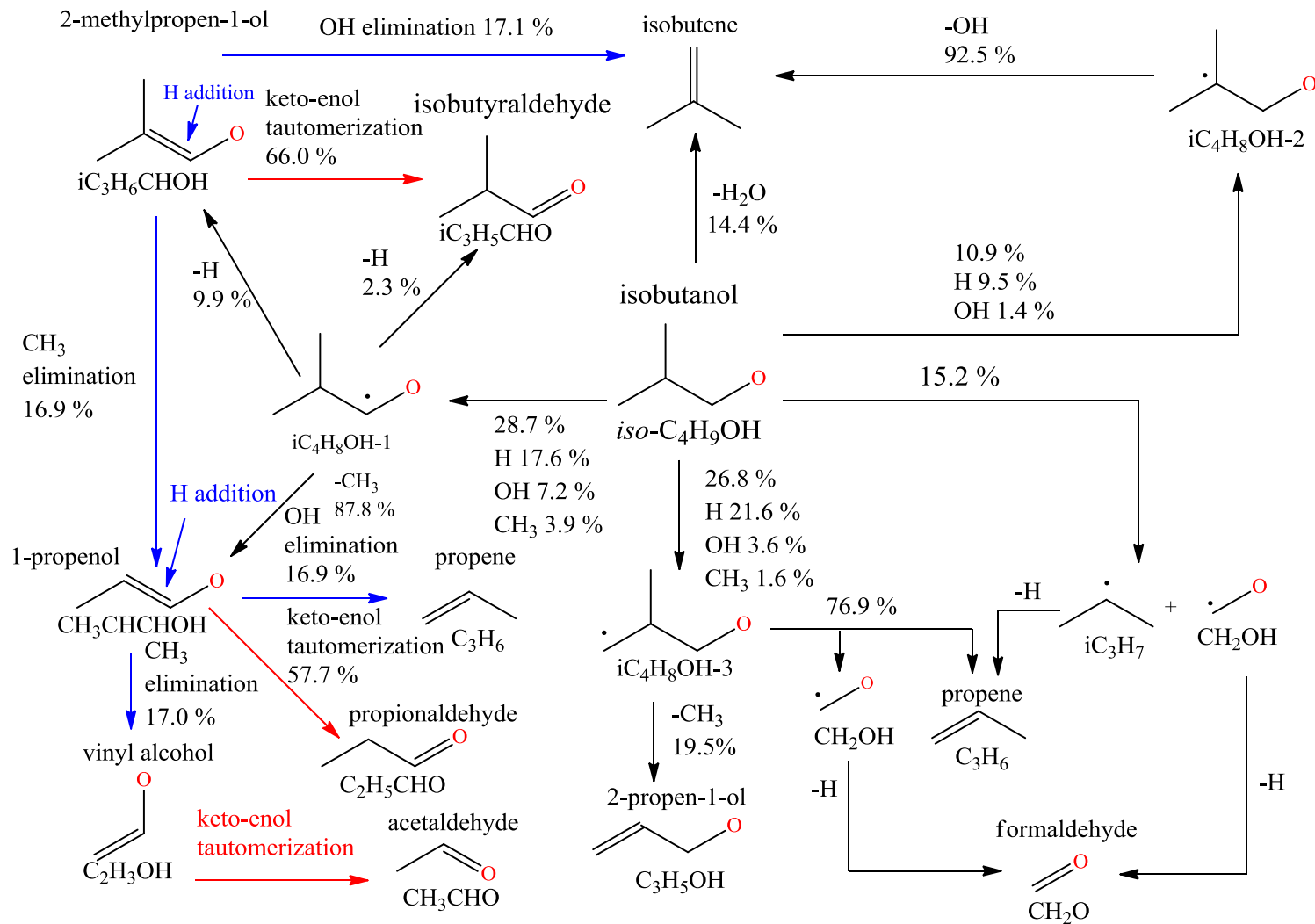
Normalized species concentration profiles for the pyrolysis of 1.5% *iso*-C₄H₉OH diluted in Ar, effective heating time = 2.6–3.0 ms, lines simulation.

tert-Butanol



Normalized species concentration profiles for the pyrolysis of 1.0% *tert*-C₄H₉OH in Ar, effective heating time = 1.5 – 1.7 ms, lines simulation.

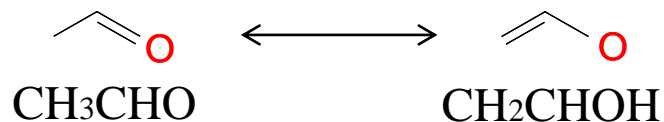
Reaction path analysis



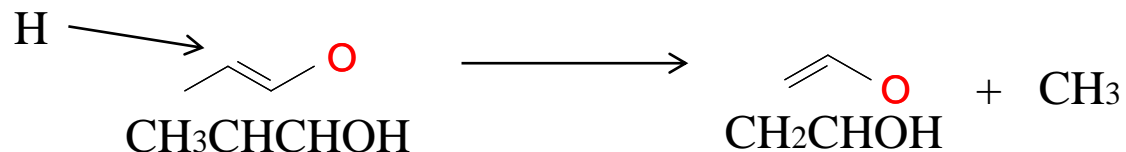
Reaction path analysis for *iso*-C₄H₉OH pyrolysis. Shock conditions: 1.5% *iso*-C₄H₉OH diluted in Ar, 1300 K, 1.9 atm, 20 % consumption

Summary

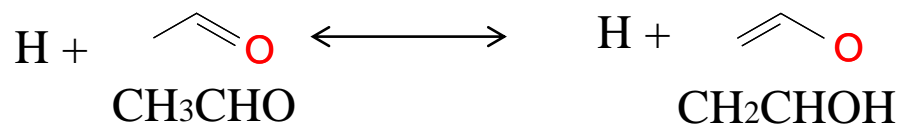
Keto-enol tautomerization



Hydrogen addition and methyl elimination



Chemically-activated H-atom (HO_2 radical and formic acid) assisted tautomerization reaction



Further reactions